Additions and Corrections

Malformin C, a New Metabolite of *Aspergillus niger* [*J. Am. Chem. Soc.*, **98**, 3365 (1976)]. By ROBERT J. ANDEREGG, KLAUS BIEMANN,* GEORGE BÜCHI,* and MARK CUSHMAN, Department of Chemistry, Massachusetts Institute of Technology, Cambridge, Massachusetts 02139.

The mass spectrum shown in Figure 4 is that of the Val-Leu-Leu derivative depicted in the figure rather than that of Leu-Leu-Ala, as stated in the legend.

The Reduction of Molecular Nitrogen, Organic Substrates, and Protons by Vanadium(II) [*J. Am. Chem. Soc.*, **98**, 7289 (1976)]. By S. I. ZONES, T. M. VICKREY, J. G. PALMER, and G. N. SCHRAUZER,* Department of Chemistry, University of California at San Diego, Revelle College, La Jolla, California 92093.

The abscissa in Figure 8 should read "μmoles $V(OH)_2$ per *400* μmoles ZrO_2·aq". In the Experimental Section, beginning with "Nitrogen Reduction with $V(OH)_2ZrO_2$·aq Systems" (p 7295), line 3 from top: "In typical experiments, *2.0* ml of this solution was injected . . . , etc." In the original version, the amounts of ZrO_2·aq were erroneously quoted to be 80 μmoles instead of the 400 μmoles actually used.

The Trans-Influence and Axial Interactions in Low Spin, Tetragonal Cobalt(II) Complexes Containing Macrocyclic and/or Cyano Ligands. Pulse Radiolytic Studies in Fluid Solution, Electron Paramagnetic Resonance Spectra at 77 K, and Single-Crystal X-Ray Structures [*J. Am. Chem. Soc.*, **99**, 429 (1977)]. By JOHN F. ENDICOTT,* J. LILIE, J. M. KUSZAJ, B. S. RAMASWAMY, WILLIAM G. SCHMONSEES, M. G. SIMIC,* MILTON D. GLICK,* and D. PAUL RILLEMA, Department of Chemistry, Wayne State University, Detroit, Michigan 48202, and the Hahn-Meitner-Institute fur Kernforschung, Berlin, GmbH, 1 Berlin 39, West Germany.

Decay times (τ_i) listed in Table III are in units of microseconds (μs)".

Isolation and Chemical Conversions of Prostaglandins from *Plexaura homomalla*: Preparation of Prostaglandin E_2, Prostaglandin $F_{2\alpha}$, and Their 5,6-Trans Isomers [*J. Am. Chem. Soc.*, **99**, 1222 (1977)]. By WILLIAM P. SCHNEIDER,* GORDON L. BUNDY, FRANK H. LINCOLN, EDWARD G. DANIELS, and JOHN E. PIKE, The Upjohn Company, Kalamazoo, Michigan 49001.

On page 1224, Figure 3 should be labeled "CD curves" instead of "ORD curves". Line 8 of the second column of the same page should read "The CD curves, Figure 3", instead of "The ORD curves . . . ".

Synthesis and Reactions of the Highly Mutagenic 7,8-Diol 9,10-Epoxides of the Carcinogen Benzo[*a*]pyrene [*J. Am. Chem. Soc.*, **99**, 1604 (1977)]. By H. YAGI, D. R. THAKKER, O. HERNANDEZ, M. KOREEDA, and D. M. JERINA,* Laboratory of Chemistry, National Institute of Arthritis, Metabolism, and Digestive Diseases, National Institutes of Health, Bethesda, Maryland 20014.

Add to footnote 1: Work was done while on leave from the Department of Chemistry, The Johns Hopkins University, Baltimore, Maryland 21218.

Kinetics of the Electron Transfer Reactions of Azaviolene Radical Ions. 2. Correlation with the Marcus Theory. The Question of Concerted Acid–Base Catalysis [*J. Am. Chem. Soc.*, **99**, 2214 (1977)]. By CLAUDE F. BERNASCONI* and HSIEN-CHANG WANG, Thimann Laboratories of the University of California, Santa Cruz, California 95064.

On page 2218, second column, third paragraph, line 5, the viscosity of AN–H_2O is η = 0.008 P instead of 0.08 P.

A New Mode of Carbonyl Scrambling. Structure and Dynamics of (1,2-Diazine)heptacarbonyldiiron(*Fe–Fe*) [*J. Am. Chem. Soc.*, **99**, 3293 (1977)]. By F. ALBERT COTTON,* BRIAN E. HANSON, JACKIE D. JAMERSON, and B. RAY STULTS, Department of Chemistry, Texas A&M University, College Station, Texas 77843.

The critical features of Figure 2 as printed are illegible. A clearer copy of the figure is shown below.

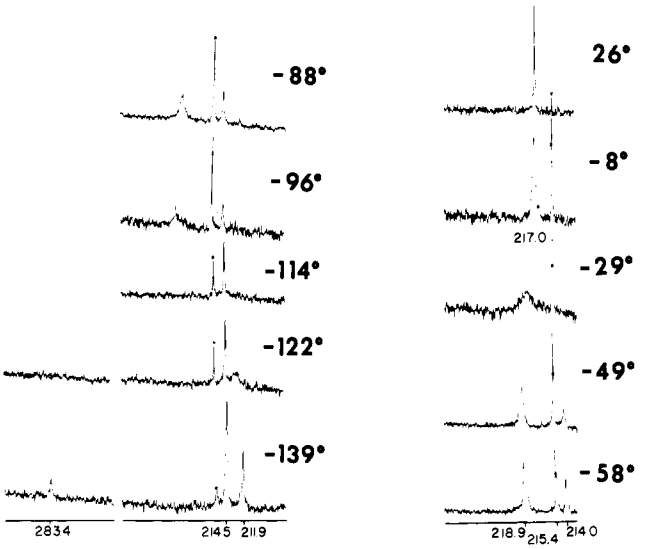

Figure 2. Carbon-13 NMR spectra in the carbonyl region at various temperatures. The peak due to the impurity $(C_4H_4N_2)Fe(CO)_4$ is marked with an asterisk. Chemical shifts are in parts per million downfield from Me_4Si.